# Assessment and interpretation of internal doses : uncertainty and variability

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This presentation has neither been approved nor endorsed by the Main Commission of ICRI

# Internal exposures are managed by the use of the committed effective dose

$$e(\tau) = \sum_{T} w_{T} \left[ \frac{h_{T}^{M}(\tau) + h_{T}^{F}(\tau)}{2} \right]$$

## Calculating committed effective dose after internal contamination is a complex procedure





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# Generic biokinetic model









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**Complex procedure, limited to experts** 

# ICRP has defined concepts and tools, to allow non-specialist to perform dose assessment



## 1. Dose per unit intake

## 2. Retention functions







	t <sub>1/2</sub>	Effective dose coefficients (Sv Bq <sup>-1</sup> )						
Nuclide		Inhalation. $e_{inh}(50)$				Ingestion		
		Туре	$f_1$	1μmAMAD	5µmAMAD	$f_1$	$e_{ing}(50)$	
a-47	4.53d	м	0.300	1.8E-09	2.1E-09	0.300	1.6E-09	
candium								
c-43	3.89h	S	1.0E-04	1.2E-10	1.8E-10	1.0E-04	1.9E-10	
c-44	3.93h	s	1.0E-04	1.9E-10	3.0E-10	1.0E-04	3.5E-10	
c-44m	2.44d	S	1.0E-04	1.5E-09	2.0E-09	1.0E-04	2.4E-0	
c-46	83.8d	s	1.0E-04	6.4E-09	4.8E-09	1.0E-04	1.5E-0	
c-47	3.35d	s	1.0E-04	7.0E-10	7.3E-10	1.0E-04	5.4E-1	
c-48	1.82d	s	1.0E-04	1.1E-09	1.6E-09	1.0E-04	1.7E-0	
c-49	0,956h	s	1.0E-04	4.1E-11	6.1E-11	1.0E-04	8,2E-1	
itanium								
i-44	47.3y	F M S	0.010 0.010 0.010	6.1E-08 4.0E-08 1.2E-07	7.2E-08 2.7E-08 6.2E-08	0.010	5.8E-09	
i-45	3.08h	F M S	0.010 0.010 0.010	4.6E-11 9.1E-11 9.6E-11	8.3E-11 1.4E-10 1.5E-10	0.010	1.5E-10	





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#### In this case, intake = Bq measured x 10<sup>5</sup>

## **Retention functions**



Fig. A.10.3. <sup>234</sup>U Inhalation Type S: predicted values (Bq per Bq intake) following acute intake.



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## Main ICRP publications on these topics

#### **For workers**

#### Publication 30 (ICRP, 1979, 1980, 1981, 1988)

dose coefficients and ALI for inhalation and ingestion. based on Reference man (Publication 23, 1975) and 1977 recommendations (Publication 26, 1977).

#### Publication 68 (ICRP, 1994)

updated dose coefficients following 1991 Recommendations (Publication 60, 1991), HRTM (Publication 66, 1994), new skeletal data (Publication 70, 1995) and revised systemic biokinetic models. No ALI anymore.

#### Publications 54 and 78 (ICRP, 1988, 1997)

guidance on the design of monitoring programs and the interpretation of results, to estimate doses to workers following radionuclide inhalation or ingestion. Provide predicted values of measured quantities after intake.



### Main ICRP publications on these topics

#### For the members of the public

#### Publications 56, 67, 69, 71 and 72 (ICRP, 1989, 1993, 1995)

age-specific dose coefficients for inhalation and ingestion for 91 elements, using up-to-date models and latest ICRP recommendations.

Publications 88 and 95 (2001,2004) Dose to embryo/fetus and infants



### **Progress and changes made during this period**

#### In physiology and biokinetic models

- New data on Reference man (ICRP 89, 2002)
- Human Alimentary Tract Model (ICRP 100, 2006)



#### The Human alimentary tract model

#### The former model



#### The Human alimentary tract model



### **Progress and changes made during this period**

#### In physiology and biokinetic models

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- New element specific systemic models, physiologically realistic

## Systemic model for lodine



The former model (ICRP 1994, 1997)

The new model

#### Three subsystems:

- circulating inorganic iodide;
- thyroidal organic iodine
- extrathyroidal organic iodine.



#### Systemic model for Strontium



**Figure 10-1. Structure of the biokinetic model for systemic strontium.** Abbreviations: exch = exchangeable, nonexch = non-exchangeable

#### The former model (ICRP 1989)

#### The new model



### Systemic model for Radon



#### Figure 12.6. Structure of the biokinetic model for systemic radon.

Abbreviations: RT-air = respiratory tract air; Blood-A = arterial blood; Blood-V = venous blood; Breast-g = glandular breast tissue; Breast-a = adipose tissue in breast; HATM = Human Alimentary Tract Model (ICRP, 2006).

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- New data on Reference man (ICRP 89, 2002)
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# Particle transport model (ICRP 66 HRTM)



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# Particle transport model (ICRP 103)



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Compound

Default Type F

Absorption parameter valuesType $f_r$  $s_r (d^{-1})$  $s_s (d^{-1})$ **1.0100** 

#### Compound

#### Default Type F

Uranyl nitrate, UO<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub> U-Tri-butyl-phosphate Uranium peroxide hydrate

Absorp	Туре		
<i>f</i> <sub>r</sub>	$s_{r}(d^{-1})$	$s_{s} (d^{-1})$	
1.0	100		
0.9	3	0.005	(F)
0.97	12	0.002	(F)
0.9	0.9	0.024	(F)



#### Compound Absorption parameter values Type $f_{\rm r} = S_{\rm r} ({\rm d}^{-1}) = S_{\rm s} ({\rm d}^{-1})$ 1.0 Default Type F 100 Uranyl nitrate, $UO_2(NO_3)_2$ (F) 0.9 3 0.005 U-Tri-butyl-phosphate 0.97 12 0.002 (F) Uranium peroxide hydrate 0.9 (F) 0.9 0.024 **Default Type M** 0.1 100 0.005

#### Absorption parameter values Compound Type $s_{\rm r} ({\rm d}^{-1}) = s_{\rm s} ({\rm d}^{-1})$ *f*<sub>r</sub> Default Type F 1.0 100 Uranyl nitrate, UO<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub> (F) 0.9 3 0.005 U-Tri-butyl-phosphate (F) 0.97 12 0.002 Uranium peroxide hydrate 0.9 (F) 0.9 0.024 **Default Type M** 0.1 100 0.005 Ammonium diuranate, ADU (M) 8.0 0.7 0.02 Uranium tetrafluoride 0.6 0.15 0.005 (M) (M) Uranium trioxide 8.0 1 0.01 $U_3O_8$ 1 0.04 0.004 (M)

Compound	Absorption parameter values			
	<i>f</i> r	$s_{r} (d^{-1})$	$s_{s} (d^{-1})$	
Default Type F	1.0	100		
Uranyl nitrate, UO <sub>2</sub> (NO <sub>3</sub> ) <sub>2</sub>	0.9	3	0.005	(F)
U-Tri-butyl-phosphate	0.97	12	0.002	(F)
Uranium peroxide hydrate	0.9	0.9	0.024	(F)
Default Type M	0.1	<b>100</b>	0.005	
Ammonium diuranate, ADU	0.8	0.7	0.02	(M)
Uranium tetrafluoride	0.6	0.15	0.005	(M)
Uranium trioxide	0.8	1	0.01	(M)
$U_3O_8$	0.04	1	0.004	(M)
Default Type S	0.001	<b>100</b>	0.0001	
Uranium dioxide	0.015	1	0.0005	(S)

# Default parameter values Type F, M, S





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#### In dosimetry and monitoring

- Development of adult reference computational phantom, based on the new ref man (ICRP 110, 2009)
- New skeletal dosimetry (ICRP 116, 2010)
- Revised nuclear decay data (ICRP 107, 2008)

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- Revised nuclear decay data (ICRP 107, 2008)
- Concept of dose per content



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# Progress and changes made during this period (con't)

#### **In ICRP recommendations**

- Adoption of the use of realistic phantoms (ICRP 103, 2007)
- Changes in weighting factors (ICRP 103, 2007)
- Changes in calculation of equivalent dose (ICRP 103, 2007)





# Progress and changes made during this period (con't)

#### **In ICRP recommendations**

- Adoption of the use of realistic phantoms (ICRP 103, 2007)
- Changes in weighting factors (ICRP 103, 2007)
- Changes in calculation of equivalent dose (ICRP 103, 2007)
- Need to provide dose coefficients for Radon (ICRP 2009)



# Progress and changes made during this period (con't)

These new data and recommendations supported a revision of the past reports and provision of new dose coefficients with guidance on monitoring programs and data interpretation

Done for external dosimetry (ICRP 116, 2010) Need to be done for internal dosimetry



#### **Revision of the reports on internal exposure**

**Division of the work in two parts :** 

- Revision of models and dose coefficients for workers (OIR series)

- Revision of models and dose coefficients for members of the public (Age dependant series, Embryo and fetus, maternal transfer,..)



## The OIR series 5 volumes

#### **OIR Part 1 (ICRP Publication 130)**

- Control of occupational exposures to radionuclides
- Biokinetic and dosimetric models
- Methods of individual and workplace monitoring
- Monitoring programmes
- General aspects of retrospective dose assessment

# To be published today!!

## The OIR series 5 volumes

#### **OIR Part 2 to 5**

#### For each element section:

- Chemical forms in the workplaces
- Principal radioisotopes, physical half-lives and decay modes
- Review of data on inhalation, ingestion and systemic biokinetics
- Structure of biokinetic models and parameter values
- Monitoring techniques and typical detection limits
- Dose coefficients, reference bioassays functions and dose per content functions in printed document and/or electronic annexes

## The OIR series 5 volumes

#### **OIR Part 2**

Hydrogen (H), Carbon (C), Phosphorus (P), Cobalt (Co), Zinc (Zn), Strontium (Sr), Yttriu Molybdenum (Mo) and Technetium (Tc).

OIR Part 3 Ruthenium (Ru), Antimony (Sb), Tellurium ( Iridium (Ir), Lead (Pb), Bismuth (Bi), Poloniu Thorium (Th) and Uranium (U).

#### **OIR Part 4**

Lanthanides series, actinium (Ac), protactiniu

2017

2016

suranic elements

(Rn), Radium (Ra),

alcium (Ca), Iron (Fe),

m (Zr), Niobium (Nb),

Caesium (Cs), Barium (Ba),

OIR Part 5 Fluorine (F), Sodium (Na), Magnesium (Mg), Potassium (K), Manganese (Mn), Nickel (Ni), Selenium (Se), Molybdenum (Mo), Technetium (Tc) and Silver (Ag) and most of the others INTERNATIONAL COMMISSION ON RADIOLOGICAL PROTECTION **Uncertainty and variability in dose calculation** 

# Uncertainty : lack of knowledge of a central value for a population

## Variability : difference between members result from physiological or environmental factors

# Variability may induce uncertainty !!



**Uncertainty in dose calculation** 

**Depends on uncertainties :** 

- in measurement of activity
- in exposure scenario (route, time, RN, form,..)
- in biokinetic and dosimetric models



### on parameter values AND on model structure



# Depends on available information to derive biokinetic models

- Human data (available for Essent. elts + Cs, Pb, Ra, U, Am, Pu,..)



- Human data
- Human data on chemically similar elements (Ln)



- Human data
- Human data on chemically similar elements (Ln)

Ca, Sr, Ba, Ra chemically and physiologically similar. but Na and K chemically similar but physiologically different Cs and K chemically similar but different kinetics



# **Uncertainties in biokinetic models** - Human data - Human data on chemically similar elements - Animal data **Need extrapolation**



- Human data
- Human data on chemically similar elements
- Animal data
- Animal data AND chemically similar elements



There are many sources of uncertainty but :

Reference models and parameter values are fixed by convention and are not subject to uncertainty

They are intended to be used for optimisation and demonstration of compliance with dose limits

In case where models are used for other scientific purposes, uncertainties and variability may be taken into account.



## Conclusions

Determination of internal doses is a complex procedure

Many tools available to directly assess dose from activity measurements

Based on the use of biokinetic and dosimetric models

No uncertainty for calculation of effective dose



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